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IN THE CLAIMS

1. (Currently Amended) A method of manufacturing zirconia-alumina body, comprising:

mixing monoclinic zirconia, yttria, and alumina with a at least one solvent to form a mixture, wherein thesaid zirconia has about 100 ppm silica or less;

drying said mixture to form a dried mixture;

disposing said dried mixture adjacent to an unfired alumina body; and

co-firing said dried mixture and said unfired alumina body to form a zirconia-alumina body, wherein said zirconia-alumina body comprises about 1 weight% to about 45 weight% monoclinic phase zirconia, based upon the total weight of thesaid zirconia-alumina body.

2. (Currently Amended) The method of manufacturing zirconia-alumina body of Claim 1, further comprising mixing a at least one dispersant into thesaid mixture, and wherein thesaid zirconia-alumina body comprises about 15 weight% to about 30 weight% monoclinic phase zirconia, with the balance cubic and tetragonal phases, based upon the total weight of the zirconia.

3. (Original) The method of manufacturing zirconia-alumina body of Claim 2, wherein said dispersant is selected from the group consisting of phosphate ester, Menhaden fish oil, sulfosuccinate, castor oil, and mixtures comprising at least one of the foregoing.

4. (Currently Amended) The method of manufacturing zirconia-alumina body of Claim 1, further comprising adding a at least one binder and a at least one plasticizer to said mixture.

5. (Original) The method of manufacturing zirconia-alumina body of Claim 4, further comprises de-airing said mixture.

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6. (Currently Amended) The method of manufacturing zirconia-alumina body of Claim 1, wherein thesaid zirconia-alumina body comprises about 18 weight% to about 25 weight% monoclinic phase zirconia, ~~with the balance cubic and tetragonal phases, based upon the total weight of the zirconia.~~

7. (Currently Amended) The method of manufacturing zirconia-alumina body of Claim 4, wherein said ~~at least one~~ binder is selected from the group consisting of polyvinyl butyral, poly methyl methacrylate, polyvinyl formal, and mixtures comprising of at least one of the foregoing.

8. (Currently Amended) The method of manufacturing zirconia-alumina body Claim 4, wherein said ~~at least one~~ plasticizer is selected from the group consisting of butyl benzyl phthalate, glycols, phthalates, and mixtures comprising at least one of the foregoing.

9. (Previously Presented) The method of manufacturing zirconia-alumina body of Claim 1, wherein said mixture and said alumina body have a sintering mismatch of less than about 5%.

10. (Original) The method of manufacturing zirconia-alumina body of Claim 1, wherein said co-firing is performed at a temperature about 1,375°C to about 1,550°C.

11. (Original) The method of manufacturing zirconia-alumina body of Claim 10, wherein said co-firing is performed at a temperature of about 1,500°C to about 1,530°C.

12. (Currently Amended) The method of manufacturing zirconia-alumina body of Claim 1, wherein said ~~at least one~~ solvent is selected from the group consisting of xylene, ethanol, and mixtures comprising at least one of the foregoing.

13. (Currently Amended) The method of manufacturing zirconia-alumina body of Claim 1, wherein thesaid zirconia-alumina body comprises up to about 95 mole% zirconia, up to about 10 mole% yttrium oxide, and up to about 10 mole% alumina, based upon thesaid total weight of thesaid zirconia-alumina body.

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14. (Currently Amended) The method of manufacturing zirconia-alumina body of Claim 13, wherein thesaid zirconia-alumina body comprises about 85 to about 93 mole% zirconia, about 3 to about 7 mole% yttrium oxide, and about 3 to about 7 mole% alumina, based upon saidthe total weight of said the zirconia-alumina body.

15. (Currently Amended) The method of manufacturing zirconia-alumina body of Claim 1, further comprising metallizing thesaid unfired zirconia body to form an electrode on a first side and a second side of said zirconia body.

16. (Currently Amended) A method of manufacturing a sensor, comprising:
mixing monoclinic zirconia, yttria, and alumina with a at least one solvent to form a mixture, wherein said zirconia has about 100 ppm silica or less;
drying said mixture to form an unfired zirconia body;
disposing an electrode on each side of said unfired zirconia body;
connecting each electrode to an electrical lead;
disposing said unfired zirconia body adjacent to an unfired alumina body to form an unfired zirconia-alumina body, wherein one of said electrodes is disposed between said zirconia body and said alumina body; and
co-firing said unfired zirconia-alumina body to form a co-fired zirconia-alumina body comprising about 1 weight% to about 45 weight% monoclinic phase zirconia, based upon thesaid total weight of thesaid zirconia-alumina body.

17. (Original) A method of manufacturing a sensor as in Claim 16, further comprising disposing a protective layer adjacent to said unfired zirconia body on a side opposite said unfired alumina body.

18. (Original) A method of manufacturing a sensor as in Claim 16, further comprising disposing support layers adjacent to said unfired alumina body, with a heater disposed within said support layers.

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19. (Original) A method of manufacturing a sensor as in Claim 18, further comprising disposing a ground plane in said support layers, between said heater and said alumina body.

20. (Currently Amended) The method of manufacturing a sensor as in Claim 16, wherein ~~thesaid~~ zirconia-alumina body comprises about 15 weight% to about 30 weight% monoclinic phase zirconia, ~~with the balance cubic and tetragonal phases, based upon the total weight of the zirconia.~~

21. (Currently Amended) The method of manufacturing a sensor as in Claim 16, wherein ~~thesaid~~ zirconia-alumina body comprises about 18 weight% to about 25 weight% monoclinic phase zirconia, ~~with the balance cubic and tetragonal phases, based upon the total weight of the zirconia.~~

22. (Currently Amended) The method of manufacturing a sensor as in Claim 16, further comprising adding a at least one binder and a at least one plasticizer to said mixture.

23. (Previously Presented) The method of manufacturing a sensor as in Claim 22, further comprises de-airing said mixture.

24. (Currently Amended) The method of manufacturing a sensor as in Claim 22, further comprising a at least one dispersant selected from the group consisting of phosphate ester, Menhaden fish oil, sulfosuccinate, castor oil, and mixtures comprising at least one of the foregoing.

25. (Currently Amended) The method of manufacturing a sensor as in Claim 22, wherein said at least one binder is selected from the group consisting of polyvinyl butyral, poly methyl methacrylate, poly vinyl formal, and mixtures comprising of at least one of the foregoing.

26. (Currently Amended) The method of manufacturing a sensor as in Claim 22, wherein said at least one plasticizer is selected from the group consisting of butyl benzyl phthalate, glycols, phthalates, and mixtures comprising at least one of the foregoing.

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27. (Previously Presented) The method of manufacturing a sensor as in Claim 16, wherein said mixture and said alumina body have a sintering mismatch of less than about 5%.

28. (Previously Presented) The method of manufacturing a sensor as in Claim 16, wherein said co-firing is performed at a temperature about 1,375°C to about 1,550°C.

29. (Previously Presented) The method of manufacturing a sensor as in Claim 28, wherein said co-firing is performed at a temperature of about 1,500°C to about 1,530°C.

30. (Currently Amended) The method of manufacturing a sensor as in Claim 16, wherein said ~~at least one~~ solvent is selected from the group consisting of xylene, ethanol, and mixtures comprising at least one of the foregoing.

31. (Currently Amended) The method of manufacturing a sensor as in Claim 16, wherein ~~thesaid~~ zirconia-alumina body comprises up to about 95 mole% zirconia, up to about 10 mole% yttrium oxide, and up to about 10 mole% alumina, based upon ~~thesaid~~ total weight of ~~thesaid~~ zirconia-alumina body.

32. (Currently Amended) The method of manufacturing a sensor as in Claim 31, wherein ~~thesaid~~ zirconia-alumina body comprises about 85 mole% to about 93 mole% zirconia, about 3 mole% to about 7 mole% yttrium oxide, and about 3 mole% to about 7 mole% alumina, based upon ~~thesaid~~ total weight of ~~thesaid~~ zirconia-alumina body.

33 - 35. (Cancelled)

36. (Previously Presented) The method of manufacturing zirconia-alumina body of Claim 1, wherein said zirconia has a total impurity amount of less than about 1,000 parts per million.

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37. (Previously Presented) The method of manufacturing a sensor as in Claim 16, wherein said zirconia has a total impurity amount of less than about 1,000 parts per million.

38. (Currently Amended) A method of manufacturing zirconia-alumina body, comprising:

mixing yttria stabilized zirconia, with monoclinic phase zirconia, yttria, and alumina with a at least one solvent to form a mixture;

drying said mixture to form a dried mixture;

disposing said dried mixture adjacent to an unfired alumina body; and

co-firing said dried mixture and said unfired alumina body to form a zirconia-alumina body, wherein said zirconia-alumina body comprises about 1 weight% to about 45 weight% monoclinic phase zirconia, based upon the total weight of the said zirconia-alumina body.

39. (Currently Amended) The method of manufacturing zirconia-alumina body of Claim 38, wherein the said zirconia-alumina body comprises about 15 weight% to about 30 weight% monoclinic phase zirconia, with the balance cubic and tetragonal phases, based upon the total weight of the zirconia.

40. (Currently Amended) The method of manufacturing zirconia-alumina body of Claim 39, wherein said the zirconia-alumina body comprises about 18 weight% to about 25 weight% monoclinic phase zirconia, with the balance cubic and tetragonal phases, based upon the total weight of the zirconia.

41. (Currently Amended) The method of manufacturing zirconia-alumina body of Claim 38, wherein the said zirconia-alumina body comprises about 85 to about 93 mole% zirconia, about 3 to about 7 mole% yttrium oxide, and about 3 to about 7 mole% alumina, based upon the said total weight of the said zirconia-alumina body.

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42. (Currently Amended) A method of manufacturing zirconia-alumina body, comprising:

mixing yttria stabilized zirconia, yttria, and alumina with a at least one solvent to form a mixture;

drying said mixture to form a dried mixture;

disposing said dried mixture adjacent to an unfired alumina body; and

co-firing said dried mixture and said unfired alumina body to form a zirconia-alumina body.

43. (Currently Amended) The method of manufacturing zirconia-alumina body of Claim 554, wherein said the zirconia has a total impurity amount of less than about 1,000 ppm, or lower total impurities.

44. (Currently Amended) The method of manufacturing zirconia-alumina body of Claim 43, wherein said the impurities are selected from the group consisting of silica, sodium, calcium, magnesium, iron, titanium, and chlorine.

45. (Currently Amended) The method of manufacturing zirconia-alumina body of Claim 44, wherein the said zirconia has about 100 ppm or less of each of silica, sodium, calcium, magnesium, iron, titanium, and chlorine.

46-47. (Cancelled)

48. (Currently Amended) The method of manufacturing a sensor as in Claim 16, wherein the said zirconia comprises about 1,000 ppm or lower total impurities, and wherein at least one of said the electrodes has a resistivity of about 10 ohm-cm or lower at 800°C in air.

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49. (Currently Amended) The method of manufacturing a sensor as in Claim 48, wherein said the impurities are selected from the group consisting of silica, sodium, calcium, magnesium, iron, titanium, and chlorine.

50. (Currently Amended) The method of manufacturing a sensor as in Claim 49, wherein said the zirconia comprises about 100 ppm or less of each of silica, sodium, calcium, magnesium, iron, titanium, and chlorine.

51. (Currently Amended) A method of manufacturing a sensor, comprising:
mixing yttria stabilized zirconia, monoclinic phase zirconia, and alumina with a at least one solvent to form a mixture;
drying said mixture to form an unfired zirconia body;
disposing an electrode on each side of said unfired zirconia body;
connecting each electrode to an electrical lead;
disposing said unfired zirconia body adjacent to an unfired alumina body to form an unfired zirconia-alumina body, wherein one of said electrodes is disposed between said zirconia body and said alumina body; and
co-firing said unfired zirconia-alumina body to form a co-fired zirconia-alumina body comprising about 1 weight% to about 45 weight% monoclinic phase zirconia, based upon a the total weight of said the zirconia-alumina body.

52. (Currently Amended) The method of manufacturing a sensor as in Claim 51, wherein said the zirconia comprises about 1,000 ppm or lower total impurities, and wherein at least one of said the electrodes has a resistivity of about 10 ohm-cm or lower at 800°C in air.

53. (Currently Amended) The method of manufacturing a sensor as in Claim 52, wherein said the impurities are selected from the group consisting of silica, sodium, calcium, magnesium, iron, titanium, and chlorine.

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54. (Currently Amended) The method of manufacturing a sensor as in Claim 53, wherein said the-zirconia comprises about 100 ppm or less of each of silica, sodium, calcium, magnesium, iron, titanium, and chlorine.

55. (New) A method of manufacturing zirconia-alumina body, comprising: mixing monoclinic zirconia, yttria, and alumina with a solvent to form a mixture; drying said mixture to form a dried mixture; disposing said dried mixture adjacent to an unfired alumina body; and co-firing said dried mixture and said unfired alumina body to form a zirconia-alumina body, wherein said zirconia-alumina body comprises about 1 weight% to about 45 weight% monoclinic phase zirconia, based upon a total weight of said zirconia-alumina body.